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Chemical Bonding

The two main types of chemical bonds are covalent bonds and van der Waals forces.

Covalent bond and Ionic bond Unit 2

CHAPTER 6 Chemistry in Biology. 13 Name

Date Class In your textbook, read about reactants and products. Fill in the blanks with the correct number of molecules to balance the chemical equation. Respond to each statement.

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Chapter 6 Section 2
Chemical Reactions
Answer Key

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Questions), Chemical
Bonds (Mrs. Sample).

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Workbook

Questions),

Chemical Bonds ...

CHAPTER 6 Section 2:

Chemical Reactions

Class In your textbook,

read about reactants

and products. Fill in the

blanks with the correct

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Section 2

number of molecules to balance the chemical equation. $C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O$

(1) Respond to each statement. $CO_2 + H_2O$

(3) (2) 4. State the principle that explains why there must be the same number of atoms of

**Sauquoit Valley
Central School
District / Homepage**

Section 1 Introduction
to Chapter 6 Chemical
Bonding Bonding

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Chemical Bonding

between Electroneg.
More-neg-sulfur and
difference Bond type
ative atom hydrogen
2.5 -2.1 = 0.4 polar-
covalent sulfur cesium
2.5 -0.7 = 1.8 ionic
sulfur chlorine 3.0 -2.5
= 0.5 polar-covalent
chlorine Chemical
Bonding, continued

Chapter 6 Chemical Bonding Table of Contents

Chapter 6 Chemical
Composition 1. 100

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Chemical Bonding

washers $0.110 \text{ g} \times 1$
washer = 11.0 g
(assuming 100 washers
is exact) $100. \text{ g} \times 1$
washer $0.110 \text{ g} = 909$
washers $2.500. \text{ g} \times 1$
cork $1.63 \text{ g} = 306.7 =$
 307 corks $500. \text{ g} \times 1$
stopper $4.31 \text{ g} = 116$
stoppers $1 \text{ kg} (1000 \text{ g})$
of corks contains $(1000$
 $\text{g} \times 1 \text{ cork}$

**Chapter 6 Chemical
Composition -
Francis Howell High
School**

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CHAPTER 6 REVIEW

Chemical Bonding

SECTION 2 SHORT

ANSWER Answer the following questions in the space provided. 1.

Use the concept of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the

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6 Chemical Bonding

5. 2. 6. 3. Section 1

and 3 (page 114) 1.

Second Law of Motion.

The force necessary to move the space shuttle is equal to its mass

times its acceleration.

2. First Law of Motion.

The rock stays at rest

at the top of the hill

until the boy applies

force that sends it

rolling downhill.

Sections 2 (page 115)

1. gravity 6. attraction.

2 ...

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Section 2

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Answers - Glencoe**

Chapter 6 Section 2
Chemical Reactions.
chemical reaction.
reactant. product.
activation energy. the
process by which
atoms or groups of
atoms in substances
ar.... substance that
exists before a
chemical reaction
starts; locat....
substance formed by a
chemical reaction;

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located on the right....

Chemical Bonding

**section2 chapter 6
chemical Flashcards
and Study Sets ...**

6.0: Introduction to
Organic Chemical

Process Industry : 6.1:

Carbon Black : Final
Section - May 1983

(PDF 95K) 6.2: Adipic

Acid : Final Section -
January 1995 (PDF

88K) Errata - February
2010 editorial

corrections Table 6.2-2
was updated. The

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lb/ton number for PM
was 0.1 in the table.

**Chapter 6: Organic
Chemical Process
Industry, AP 42,
Fifth ...**

Chapter 6: THE
GOVERNOR,
LIEUTENANT
GOVERNOR AND
COUNCIL, CERTAIN
OFFICERS UNDER THE
GOVERNOR AND
COUNCIL, AND STATE
LIBRARY Section 1

Salary of governor;

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Section 2

housing expenses;

other sources of

income; Section 2

Salary of lieutenant

governor; other

sources of income;

Section 3 Salary of

council members; term

limitation; Section 4

Traveling expenses of

lieutenant governor

and council members;

term ...

Chapter 6

SECTION 1 Introduction

to Chemical Bonding

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Section 2 Covalent
Bonding and Molecular

Compounds SECTION 3

Ionic Bonding and Ionic

Compounds SECTION 4

Metallic Bonding

SECTION 5 Molecular

Geometry Why It

Matters Video

HMHSscience.com GO

ONLINE Chemical

Bonding BIG IDEA

Atoms form chemical

bonds by sharing or

transferring electrons.

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Section 2

CorrectionKey=NL-A

DO NOT

**EDIT--Changes must
be made ...**

Chapter 6 Reactants
and Products □A

chemical reaction is
the process by which
atoms or groups of
atoms in substances
are reorganized into
different substances.

6.2 Chemical Reactions

□Clues that a chemical
reaction has taken
place include the
production of heat or

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Section 2

light, and formation of
a gas, liquid, or solid.

Chemical Bonding

Chapter 6 Chemistry in Biology - Hall High School

CHAPTER 6 Chemical
Bonding SECTION 1

Introduction to

Chemical Bonding

OBJECTIVES 1. Define

Chemical bond. 2.

Explain why most

atoms form chemical

bonds. 3. Describe

ionic and covalent

bonding. 4. Explain

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why most chemical bonding is neither

purely ionic or purely

5. Classify bonding

type according to

electronegativity

differences.

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CHAPTER 6 Section

2:Chemical Reactions

Match the description

in Column A with the

term in Column B.

Column A Column B 7.

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Name Date Class **Chemical Bonding**

Chapter 6: Chemical Bonding Section 1- Introduction to Chemical Bonding Objectives: define chemical bond; differentiate between covalent and ionic bonding; explain why bonding occurs; use the difference in electronegativity to determine whether a bond is polar covalent, nonpolar covalent or

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Section 2
ionic
Chemical Bonding
Ch 6 -

HonorsChemWins

Introduction to
Balancing Chemical
Equations - Duration: ...
12:23. Chapter 6
Section 4: The Building
Blocks of Life -
Duration: 13:32. Mrs.
Di Stefano's Science
Presentations 2,419
views.

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